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emphasis on skills  
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A functional group

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is an atom or a group of atoms that are responsible for the physical and chemical properties of the molecule. Knowing the functional groups is a must in organic chemistry and in this post, we will go over the structure and

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interesting applications of the most common functional groups in organic chemistry.

[Functional Groups in Organic Chemistry - Chemistry Steps](#)

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chemistry is the study of compounds of carbon. Although many organic compounds are isolated from plant and animal sources, most of them can be synthesized. All the organic compounds contain the element

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carbon. Compounds of carbon with hydrogen are called hydrocarbons.

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Chemistry and Organic Chemistry students. It's comparable to MolyMod kits but the parts are harder to take apart with MolyMod. This Old Nobby kit has more 5-hole (trigonal bipyramid) and 6-hole (octahedral) atoms than you'll

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ever use and removing most of them will make the lid for the box fit ...

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General Organic  
Chemistry (GOC)  
covers the  
fundamental

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concepts in the vast field of organic chemistry. A strong understanding of the concepts that come under GOC plays a vital role in the study of relatively advanced topics (such as the mechanisms of named reactions).

[General Organic](#)

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[Chemistry \(GOC\)](#)

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Organic chemistry is the study of the chemistry of carbon compounds. A functional group is a specific structural arrangement of atoms or bonds that imparts a

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characteristic  
chemical reactivity  
to the molecule;  
alcohol group and  
carboxylic group  
(answers will vary).

[1.1. Introduction to  
organic chemistry |  
Organic Chemistry](#)

[...](#)

Organic Chemistry  
– Specific Name ...

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## Organic

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## Solution

Fehling solution A and Fehling solution B. Fehling solution A is aqueous copper sulphate and Fehling solution B is alkaline sodium potassium tartarate (Rochelle salt). These two solutions are mixed in equal amounts before test. On

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heating an  
aldehyde with  
Fehling's reagent, a  
reddish

[Organic Chemistry](#)  
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In this section are  
stepwise  
instructions on how  
to extract an  
aqueous solution

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with an organic solvent that is less dense than water (the organic layer will be on the top). As an example, the instructions are written to extract an aqueous solution three times using 25 mL diethyl ether each time (3  $\times$  25 mL).

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\text{mL ...

Solution

[4.6: Step-by-Step  
Procedures For  
Extractions -  
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Organic chemistry.  
Dilute solutions of  
 $\text{KMnO}_4$  convert  
alkenes into diols  
(glycols). This  
behaviour is also  
used as a

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## Organic

## Chemistry

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qualitative test for the presence of double or triple bonds in a molecule, since the reaction decolorizes the initially purple permanganate solution and generates a brown precipitate ( $\text{MnO}_2$ ).

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